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Substituting
and multiplying
by the factor
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density unit

kg/km^3 , the mass of the

atmosphere is

determined to

$m_{\text{atm}} = 5.092 \times 10^{18}$

kg Discussion

Performing the

analysis with

excel would

yield exactly

the same

results. EES

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final result: a=1

.2025166 b=-0.101

67 c=0.0022375 r=6

377 h=25 m=4*pi*(a

*r^2*h+r*(2*a+b*

r)*h^2/2+(a+2*b*

r+c*r^2)*h^3/3+(

b+2*c*r)*h^4/4+c

*h^5/5)*1E+9

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pressure varies from 0.5 MPa to 1.5 MPa. The volume of the tank is to be plotted versus the initial pressure, and the results are to be discussed.

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is, a straight
line
representing
composition), we
are dealing with
a two component
system

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The results are:

$$\rho(z) = a + bz + cz^2 = 1.20252 - 0.101674z +$$

0.0022375z² for
the unit of kg/m³, (or, $\rho(z) =$

$$(1.20252 - 0.101674z + 0.0022375z^2$$

) $\times 10^9$ for the
unit of kg/km³

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) where z is the vertical distance from the earth surface at sea level. At $z = 7$ km, the equation would give $\rho = 0.60 \text{ kg/m}^3$.

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